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REENTRANT NEMATIC AND SMECTIC C PHASES IN BINARY SYSTEMS OF $4-(\beta-CYANETHYL)-PHENYL$ 4-n-ALKYLOXY-CINNAMATES

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(Submitted for publication September 25, 1980)

Abstract: In binary systems of 4-(β -cyanethyl)-phenyl 4-n-alkyloxycinnamates reentrant nematic and reentrant smectic C phases occur. In three systems studied the phase sequences SC Nre SA N Is and SC(re) Nre SC SA N Is were found.

Cladis first observed that binary mixtures of p-cyano substituted azomethines 1 and cyanoalkylbiphenyls 2 show the following phase sequence with decreasing temperature: nematic(N), smectic A (SA) and again nematic. The low temperature nematic phase was called "reentrant" nematic phase (Nre). Later on Engelen et al. 3 found reentrant nematic phases also in binary mixtures of terminal polar liquid crystals and terminal non-polar liquid crystals. We present first investigations about 4-(β -cyanethyl)-phenyl 4-n-alkyl-oxycinnamates which show a strong tendency to form reentrant phases in binary mixtures. The lower members of this homologous series exhibit nematic phases; the higher homologues possess additional smectic phases (Table 1).

We have studied the diagrams of state of binary systems by means of the contact method⁵ and by investigations of singular concentrations. Combining a nematic homologue of the series with a homologue exhibiting SC SA or SC SA N polymorphism, analogous diagrams were obtained (see Figures 1-3). The common feature of these diagrams is the concave curvature of the SA-N transition curve and the convex

Is

Table 1

$$\texttt{C}_n\texttt{H}_{2n+1}\texttt{O} - \underbrace{\texttt{C}\texttt{H} \texttt{=}\texttt{C}\texttt{H} \texttt{-}\texttt{C}\texttt{O}\texttt{O}} - \underbrace{\texttt{C}\texttt{H}_2 \texttt{-}\texttt{C}\texttt{H}_2 \texttt{-}\texttt{C}\texttt{N}}$$

n	Cr			s _C		s_{A}		N		Is
6	•	64 ^x ;	71.5	-		-		•	109.2	•
8	•	71.5 ^x ;	87.2	-		-		•	108.2	•
10	•	68 ^x ;	72	(•	58)	•	109	•	109.5	•
12	•	73.5 ^x ;	82	(•	58)	•	117	-		•
		:	Cr S _A , S _C	solid cry smectic A nematic						

The numbers between the phase symbols are the transition temperatures (°C). Brackets denote metastable phases.

isotropic liquid

*melting points of metastable solid modifications.

curvature of the S_C -N curve with respect to the temperature axis. Comparing the diagrams of state in Figures 1-3, we can derive that the existence range of smectic phases generally increases with increasing chain lengths of the components in the binary system. The complicated shape of the S_A -N and S_C -N curves gives rise to the occurrence of reentrant nematic and reentrant smectic C phases ($S_C(r_e)$).

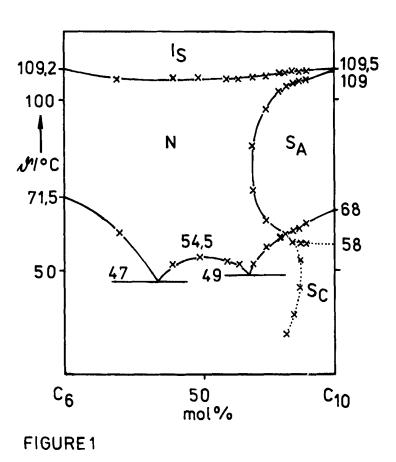
Additionally to the well known sequence N_{re} S_A N in all diagrams, the unusual sequences S_C N_{re} S_A N and $S_C(re)$ N_{re} S_C S_A N were found. The unusual phase sequences occur in a limited concentration range. The S_C range between S_A and N_{re} is relatively small (0,3-3 K). In most cases (except for Figure 3), the $S_{C(re)}$ phase could be detected only by rapid supercooling. Although the textures of the N_{re} and S_C resp. $S_{C(re)}$ phases are not very different, these phases could be distinguished by the different behaviour after mechanical and dielectric deformations. Furthermore, the transitions were clearly indicated by the appearance of a transition front. The phase transitions were examined by

differential scanning calorimetry (DSC2, Perkin-Elmer). In principle, all phase transitions found by optical observations could be detected also by calorimetry, except the transition N_{re} -SC. This is obviously caused by the fact that the SC range between N_{re} and S_A is too small. The order of magnitude of the transition enthalpies is shown in Table 2.

Table 2

Transition	S _{C(re)} /N _{re}	N_{re}/S_{A}	S _C /S _A	SA/N	N/Is
ΔH/Jmo1 ⁻¹	50	100-300	100-250	500	2800

Because the metastable SC phases of the pure substances (C9, C10, C12) crystallize quickly, the X-ray investigations were performed on an equimolar mixture of the C₉ and C₁₀ compounds in which the Sc phase could be supercooled about 10 K. According to preliminary measurements the layer spacing d in the SA as well as SC phases is distinctly larger than the average length l of the C9 and C10 molecules in its most The ratio d/l was found to be about stretched conformation. This behaviour is characteristic of those molecules which possess a strong polar head group (CN) on the one end and a non-polar aliphatic tail on the other end.6,7 special molecule structure leads to a kind of bilayer structure with interdigitated molecules in each layer. Finally, it should be noted that the 4-(β -cyanethyl)-phenyl 4-n-alkyloxycinnamates form reentrant nematic and smectic phases in binary mixtures with a great number of liquid crystals possessing quite different chemical structures. These results will be published elsewhere.



NOTE: In each of the figures, the melting curve drawn relates to that solid modification which was obtained in a large concentration range. The dotted curves indicate phase transitions in the supercooled state.

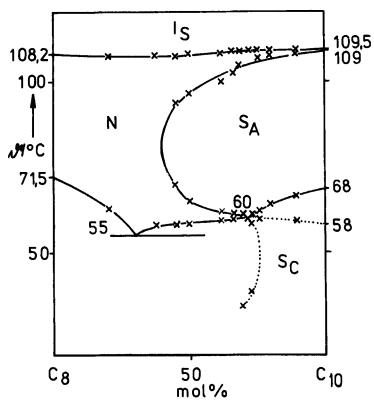
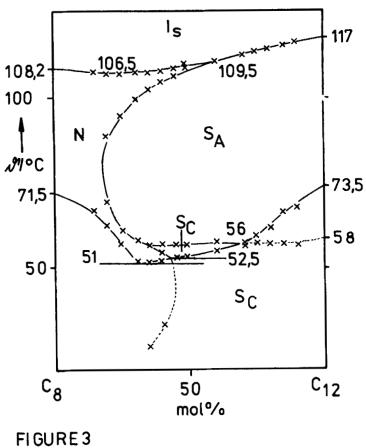


FIGURE 2



FIGURES

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